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Magnetic Organic Superconductors Based on BETS Molecules—Interplay of Conductivity and Magnetism

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BETS (=bis(ethylenedithio)tetraselenafulvalene) Among superconductors, the systems with FeX₄- (X=Cl, Br) anions exhibit salient electronic properties due to the coupling between π and d λ-BETS₂FeCl₄ undergoes successive transitions as, electron systems. antiferromagnetic insulating phase \rightarrow metallic ferromagnetically oriented Fe^{3+} spins \rightarrow superconducting phase $(H//\text{conduction plane}) \rightarrow \text{metallic phase}$, with increasing magnetic field. exhibits κ-BETS₂FeX₄ successive antiferromagnetic superconducting transitions with decreasing temperature. Resistivity anomaly observed at antiferromagnetic transition temperature indicates an evidence for the coupling between π and d electron systems

<u>Keywords:</u> antiferromagnetic organic superconductor; BETS; magnetic organic superconductor

INTRODUCTION

The discovery of the first organic superconductor (Bechgaard salt) in 1980 has opened new era of the chemistry and physics of molecular conductors^[1]. Its crystal structure gave an important hint to develop the molecular metals with two-dimensional electronic band structures^[2]. In fact, various types of molecular metals with two-dimensional cylindrical Fermi surfaces were prepared in the later half of 1980s. The typical example is κ -type organic superconductors^[3]. However, the investigation on the development of new organic superconductors seemed to be rather faded out recently because the elevation of T_c has been long stopped since the discovery of κ-(BEDT-TTF)₂ Cu[N(CN)₂]Cl ^[4]. On the other hand another stream started to grow around the middle of 1990s. That is, the development of new molecular conductors such as paramagnetic organic superconductor and ferromagnetic metal based on BEDT-TTF molecules and magnetic anions has attracted an increasing interest^[5,6]. Although the interaction between π metal electrons in the two-dimensional donor layers and the localized 3d magnetic moments of the anion sites is expected to produce bi-functional properties, it seems to be not so easy to develop the organic conductors with considerable π -d coupling.

About a decade ago, we have tried to prepare organic conductors incorporating magnetic moments with the aim of examining the possibility of developing an organic conductors such as "organic Kondo system", where π metal electrons interact with localized magnetic moments of the anions at low temperatures^[7]. We have prepared BETS conductors with magnetic anions such as FeCl₄- and FeBr₄- and also analogous nonmagnetic anions such as GaCl₄-, GaBr₄-, InCl₄- and InBr₄- (BETS = bis(ethylenedithio)tetraselena-fulvalene). Here, we report the results of the recent investigation on the electric and magnetic properties of BETS conductors with tetrahalide metal anions.

ELECTRICAL PROPERTIES OF BETS CONDUCTORS WITH TETRAHALIDE ANIONS

Crystals of BETS conductors with tetrahalide anions MX₄- (M=Fe, Ga, In, Tl; X=Cl, Br) were prepared electrochemically. There are mainly two modifications. Needle crystals with triclinic unit cells named λ -type crystals were obtained when [(C₂H₅)₄N]MCl₄ (M=Ga, Fe) were used as electrolyte. Crystals of the other modification with orthorhombic κ -type structures were obtained as plate crystals for all these MX₄⁻ anions. At first sight, the variety of λ -type salt seems to be very limited. However, due to the close similarity of the lattice constants of the GaCl₄ and FeCl₄ systems, Ga and Fe atoms can be easily exchanged to produce the alloy system, λ -(BETS)₂Fe_xGa_{1-x}Cl₄ (0<x<1) exhibiting various novel electronic properties. As reported before, the resistivity behavior of λ -(BETS)₂GaCl₄ closely resembles that of λ -(BETS)₂FeCl₄ except for low temperature region. Both salts show gradual resistivity increase with decreasing temperature and take round resistivity maxima around 100 K (see Figure 1). Then the resistivities decrease very rapidly^[8,9]. Around 6 K, λ-(BETS)₂GaCl₄ undergoes a superconducting transition. While λ-(BETS)₂FeCl₄ exhibits a coupled antiferromagnetic and metal-insulator transition around 8.5 K, which clearly shows an importance role of the π -d coupling in FeCl₄ salt. The Cl atoms of the MCl₄⁻ can be partially exchanged by Br atom. The unit cell volume is increased with increasing Br-content. Then the π electron system feels effectively "negative pressure" and the system tends to become a semiconductor. In λ-(BETS)₂GaBr_{1.5}Cl_{2.5} semiconducting behavior. But by applying "real pressure" to this Br-rich semiconducting system, the system becomes a superconductor. Around 3 kbar, the resistivity of λ -(BETS)₂GaBr_{1.5}Cl_{2.5} is constant down to 100 K and then decreases rapidly. The system shows a superconducting transition at about $10 \text{ K}^{[9]}$ (see the inset of Figure 1).

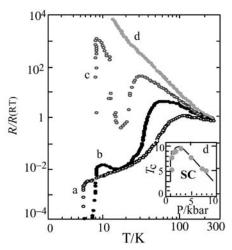


FIGURE 1 The temperature dependence of the resistivity of λ -BETS₂GaBr_xCl_{4-x}: x=0 (a), 0.5 (b), 0.7 (c), 1.5 (d). The inset is the pressure dependence of superconducting transition temperature (T_c) of λ -BETS₂GaBr_{1.5}Cl_{2.5}.

Contrary to λ -type salts, every κ -type salt is metallic at low temperatures. The systems with small size anions such as $GaCl_4$ and $FeCl_4$ exhibit smooth resistivity decrease with lowering the temperature but the systems with relatively large size anions such as $GaBr_4$ and $FeBr_4$ exhibit round resistivity maxima around 60-70 K (Tp), which is related to the "negative chemical pressure effect of Br atoms".

Recently, the electrical resistivity experiments were made on κ -(BETS)₂TlCl₄ to see the anion size effect. But all the crystals examined were destructed around 220 K. However we noticed that the resistivity measurements can be performed by coating the crystal with epoxy resin because the "effective pressure" produced by the contraction of the epoxy resin at low temperature suppresses the destruction of the crystal to some extent. The crystal coated by epoxy resin showed a superconducting transition below 2.5 K (Figure 2)^[10].

The low-temperature X-ray diffraction experiments made by using the crystal coated by epoxy resin showed the doubling of the lattice constant c around 210 K (a, b, $c \rightarrow a$, b, 2c). In Table 1, six BETS superconductors with MX₄- anions ever discovered are listed.

We have also examined BETS conductors with magnetic dianions such as $CoCl_4^{2-}$, $CoBr_4^{2-}$ and $MnBr_4^{2-}$. The 2:1 complex $(BETS)_2CoCl_4$ was insulator but the 4:1 salts with κ - and θ -type molecular arrangements were metallic down to ca. 50 K $(CoBr_4^{2-}$ and $MnBr_4^{2-})$ or 0.7 K $(CoCl_4^{2-})$. The antiferromagnetic interaction between localized magnetic moments $(Co^{2+} (S=3/2), Mn^{2+} (S=5/2))$ is very weak because the anion layers of these salts are composed of anions and crystal solvents.

TABLE 1 The temperatures characterizing the electric and magnetic properties of BETS superconductors.

	salt ^{a)}	λ-GaCl ₄	λ-FeCl ₄	κ-FeCl ₄	κ-FeBr ₄	κ-GaBr ₄	κ-TlCl ₄
	<i>T</i> _p ^{b)}	90 K	90 K	normal metal	60 K	70 K	c)
	$T_{ m N}$		8.5 K ^{d)}	0.45 K ^{e)}	2.4 K ^{e)}		
•	$T_{\rm c}$	6 K	2 K (at 3 kbar)	0.17 K	1.1 K	1 K	2 K f)

a) λ - and κ-(BETS)₂MX₄ are abbreviated to λ - and κ-MX₄, respectively.

b) The temperature corresponding to the round resistivity peak.

c) Resistivity measurements of κ -(BETS)₂TICl₄ could not be made due to the destruction of crystal around 220 K.

d) Antiferromadnetic and metal-insulator transitions take place cooperatively at 8.5 K.

e) κ -(BETS)₂FeX₄ (X=Cl, Br) undergoes a paramagnetic metal-to-antiferromagnetic metal transition.

f) Resistivity measurements were made on the crystal coated by epoxy resin.

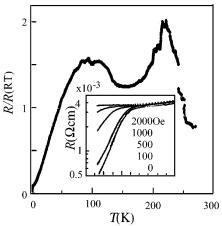


FIGURE 2 Temperature dependence of the resistivity of κ -(BETS)2TlCl4. The crystal coated by epoxy resin was used. The inset is the field dependence of the superconducting transition.

COUPLING OF π AND d ELECTRON SYSTEMS IN BETS CONDUCTORS WITH Fe $^{3+}$ IONS

As mentioned before, λ -(BETS)₂FeCl₄ undergoes π -d coupled antiferromagnetic insulating transition at 8.5 K. That is, at ambient pressure, the magnetic transition and electronic transition take place cooperatively at the same temperature. However, these two transitions will tend to be separated by applying pressure, because the interaction between Fe³⁺ spins will be enhanced at high pressure while the metal-insulator (MI) transition will be suppressed by pressure. In fact, the MI transition temperature is rapidly decreased and superconducting transition appears above 3 kbar (see Table 1), suggesting that π electron system of λ -type salt tends to take a superconducting ground state when π -d coupling is weakened^[11].

Similar to the case of λ -(BETS)₂GaBr_xCl_{4-x} (see Figure 1), the π electron system of λ -(BETS)₂FeBr_xCl_{4-x} tends to be semiconducting

with increasing Br-content. Therefore, the metal-insulator transition temperature ($T_{\rm MI}$) is enhanced in Br-rich system. For example, $T_{\rm MI}$ of λ -(BETS)₂FeBr_{0.7}Cl_{3.3} is about twice of λ -(BETS)₂FeCl₄ (see Figure 3). While the magnetic susceptibility indicates that Fe³⁺ spin system of λ -(BETS)₂FeBr_{0.7}Cl_{3.3} undergoes an antiferromagnetic transition around 8 K (magnetic transition temperature ($T_{\rm N}$) to be approximately independent of Br-content system)^[12]. It is well known that $T_{\rm MI}$ of λ -(BETS)₂FeCl₄ is completely suppressed above 11T, where π metal electrons and ferromagnetically oriented Fe³⁺ moments coexist. However, $T_{\rm MI}$ of λ -(BETS)₂FeBr_{0.7}Cl_{3.3} is unchanged at least up to 12 T (Figure 3). That is, contrary to the case of superconducting transition, the considerable π -d coupling is essential to realize the field-restored metallic state. Of course, $T_{\rm MI}$ is decreased at high pressure, where field-restored metallic state can be observed.

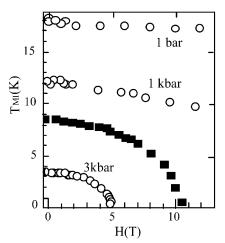


FIGURE 3 The magnetic field dependence of the metal-insulator transition temperature $(T_{\rm MI})$ of λ -(BETS)₂FeBr_{0.7}Cl_{3.3} at 1bar, 1 kbar and 3 kbar. (open circle). The closed squares are those of λ -(BETS)₂FeCl₄ at 1bar.

Recently the field-induced superconducting state (FISC) was discovered in λ -(BETS)₂FeCl₄ above 17 T^[13]. Similar phenomena were also found in λ -(BETS)₂Fe_xGa_{1-x}Cl₄. We have examined the resistivity of λ -(BETS)₂Fe_xGa_{1-x}Cl₄ ($x \approx 0.4$) which exhibits a successive metal \rightarrow superconductor \rightarrow insulator transitions with decreasing temperature. By applying magnetic field (H), the insulating ground state tended to change to a superconducting state ($H \perp b^*$ or a metallic state ($H \mid /b^*$). When H is perpendicular to b^* , the superconducting phase around zero-magnetic field is continuously changed to the "field-induced superconducting state".

In contrast to λ -(BETS)₂FeCl₄, κ -(BETS)₂FeCl₄ exhibits a smooth resistivity decrease down to very low temperatures (see Figure 4). The temperature dependence of the resistivity shows a step at 0.45 K, which agrees well with the antiferromagnetic transition temperature

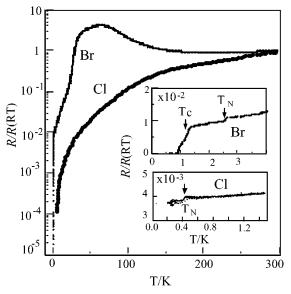


FIGURE 4 Temperature dependencies of the resistivities of κ-(BETS)₂FeCl₄ (Cl) and κ-(BETS)₂FeBr₄ (Br).

 (T_N) (see inset of Figure 4)^[14]. Magnetic susceptibility and μSR experiments revealed the superconducting transition to be 0.17 K ^[14,15]. κ -(BETS)₂FeBr₄ also shows a resistivity step at the antiferromagnetic transition temperature (2.4 K) and superconducting transition at 1.1 K ^[16]. Recent specific heat measurements suggested the 2.4 K transition of κ -(BETS)₂FeBr₄ and the 0.45 K transition of κ -(BETS)₂FeCl₄ to be pure magnetic transitions. In addition, the transition entropy of κ -(BETS)₂FeCl₄ showed the low-dimensionality of the Fe³⁺ spin system. A characteristic resistivity peak of κ -(BETS)₂FeBr₄ around 60 K was suppressed by applying pressure and the superconducting transition was decreased (e.g., 0.8 K at 2.5 kbar). These features are consistent with the concept of "negative chemical pressure effect of Br atom."

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